

Crystal Data: Hexagonal. *Point Group:* 3. In bundles of slender acicular to threadlike crystals.
Twinning: Micro-twinning observed in structure analysis.

Physical Properties: *Cleavage:* {0001}, perfect. *Tenacity:* Brittle. Hardness = 6
D(meas.) = 2.493-2.649 D(calc.) = [2.65]

Optical Properties: Transparent. *Color:* Colorless. *Luster:* Silky, brilliant.
Optical Class: Uniaxial (-). $\omega = 1.536$ $\epsilon = 1.531$

Cell Data: *Space Group:* P3. $a = 27.0597(16)$ $c = 8.5587(6)$ $Z = 54$

X-ray Powder Pattern: Vesuvius, Italy.
3.09 (100), 2.593 (30), 2.131 (25), 4.26 (15), 3.76 (12), 3.02 (12), 2.814 (12D)

Chemistry:	(1)	(2)	(3)
SiO ₂	39.2	38.87	37.99
TiO ₂		0.00	
Al ₂ O ₃	33.36	31.89	32.23
FeO		0.07	
MnO		0.00	
CaO	0.47	0.05	
Na ₂ O	2.88	0.77	
<u>K₂O</u>	<u>24.13</u>	<u>28.76</u>	<u>29.78</u>
Total	100.04	100.41	100.00

(1) Monte Somma, Italy. (2) Do.; by electron microprobe. (3) KAlSiO₄.

Polymorphism & Series: Polymorphous with kalsilite, panunzite, and trikalsilite.

Occurrence: In blocks of biotite-pyroxenite volcanic ejecta.

Association: Biotite, pyroxene, augite, melilite, calcite (Monte Somma, Italy); clinopyroxene, garnet, leucite, h  yne, latiumite (Albano, Italy).

Distribution: In Italy, from Monte Somma and Vesuvius, Campania (TL); at Albano and Marino, Colle Cimino, near Rome, Lazio.

Name: From the Greek for potassium, *kali*um, and for "friend", recognizing potassium's presence.

References: (1) Dana, E.S. (1892) Dana's system of mineralogy, (6th edition), 427. (2) Bannister, M.A. (1931) A chemical, optical, and X-ray study of nepheline and kaliophilite. *Mineral. Mag.*, 22, 569-608. (3) Smith, J.V. and O.F. Tuttle (1957) The nepheline-kalsilite system: I. X-ray data for the crystalline phases. *Amer. J. Sci.*, 255, 282-305. (4) Deer, W.A., R.A. Howie, and J. Zussman (1963) *Rock-forming minerals*, v. 4, framework silicates, 231-270. (5) Mugnaioli, E., E. Bonaccorsi, A.E. Lanza, E. Elkaim, V. Diez-G  mez, I. Sobrados, M. Gemmia, and M. Gregorkiewitz (2020) The structure of kaliophilite KAlSiO₄, a long-lasting crystallographic problem. *IUCrJ*, 7, 1070-1083.